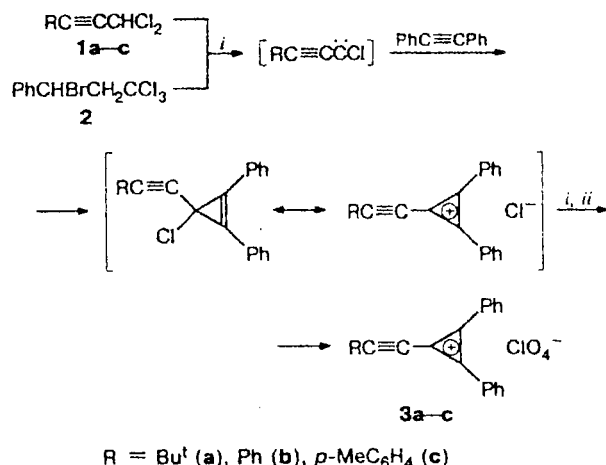


(Alk-1-ynyl)diphenylcyclopropenium perchlorates as a new class of cyclopropenium salts

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We found that generation of (alk-1-ynyl)halocarbenes by treatment of 1,1-dichloroalk-2-ynes^{1–3} (**1a–c**) or 3-bromo-1,1,1-trichloro-3-phenylpropane⁴ (**2**) with Bu^tOK in benzene at ~20 °C in the presence of a threefold excess of tolane followed by treatment of the reaction mixture, having been passed through a layer of SiO₂, with 75% perchloric acid gives (alk-1-ynyl)diphenylcyclopropenium perchlorates (**3**), previously unknown compounds of the acetylene series, as solid precipitates in 24–38% yields.



Reagents and conditions. i. Bu^tOK, benzene, 20 °C.
ii. 75% aqueous HClO₄, benzene, 20 °C.

We used this procedure to transform 1,1-dichloro-4,4-dimethylpent-2-yne (**1a**) into 1-(3,3-dimethylbut-1-ynyl)-2,3-diphenylcyclopropenium perchlorate (**3a**) in 38% yield, m.p. 70–71 °C. ¹H NMR, δ: 1.54 (s, 9 H, Bu^t); 7.86 (br.dd, 4 H, *m*-H, 2 Ph, *J*₁ = *J*₂ = 7.8 Hz); 8.04 (br.t, 2 H, *p*-H, 2 Ph, *J* = 7.8 Hz); 8.43 (br.d, 4 H, *o*-H, 2 Ph, *J* = 7.8 Hz). ¹³C NMR, δ: 28.6 (3 CH₃); 30.2 (C(CH₃)₃); 63.9 (Bu^tC≡C); 119.4 (*ipso*-C, 2 Ph); 130.4, 136.1, 139.2 (2 Ph); 138.4 (Bu^tC≡C); 144.2 (C≡C, *cyclo*-C₃⁺); 155.3 (2 CPh, *cyclo*-C₃⁺). IR, ν/cm⁻¹: 1408 (*cyclo*-C₃⁺); 2210 (C≡C).

The transformation of 1,1-dichloro-3-phenylprop-2-yne (**1b**) gave 1-(phenylethynyl)-2,3-diphenylcyclopropenium perchlorate (**3b**) in 35% yield, m.p. 66–

68 °C. ¹H NMR, δ: 7.69 (br.dd, 2 H, *m*-H, PhC≡, *J*₁ = *J*₂ = 7.5 Hz); 7.80 (br.t, 1 H, *p*-H, PhC≡, *J* = 7.5 Hz); 7.93 (br.dd, 4 H, *m*-H, 2 Ph, *J*₁ = *J*₂ = 7.7 Hz); 8.00–8.15 (m, 4 H, *p*-H, 2 Ph, *o*-H, PhC≡); 8.53 (br.d, 4 H, *o*-H, 2 Ph, *J* = 7.7 Hz). ¹³C NMR, δ: 73.5 (PhC≡C); 118.2, 119.5 (*ipso*-C, 2 Ph, PhC≡); 129.5, 134.4, 134.9 (PhC≡); 130.4, 136.2, 139.2 (2 Ph); 125.7 (PhC≡C); 142.4 (C≡C, *cyclo*-C₃⁺); 154.5 (2 CPh, *cyclo*-C₃⁺). IR, ν/cm⁻¹: 1408 (*cyclo*-C₃⁺); 2191 (C≡C).

Perchlorate **3b** was also prepared in 25% yield from 3-bromo-1,1,1-trichloro-3-phenylpropane (**2**), which, according to published data,⁴ generates (phenylethynyl)chlorocarbenes on treatment with Bu^tOK.

1,1-Dichloro-3-(*p*-tolyl)prop-2-yne (**1c**) was converted into 1-(*p*-tolylethynyl)-2,3-diphenylcyclopropenium perchlorate (**3c**) in 24% yield, m.p. 163–164 °C. ¹H NMR, δ: 2.49 (s, 3 H, Me); 7.48 (br.d, 2 H, *m*-H, *p*-Tol, *J* = 8.0 Hz); 7.91 (br.dd, 4 H, *m*-H, 2 Ph, *J*₁ = *J*₂ = 7.6 Hz); 7.92 (br.d, 2 H, *o*-H, *p*-Tol, *J* = 8.0 Hz); 8.08 (br.t, 4 H, *p*-H, 2 Ph, *J* = 7.6 Hz); 8.51 (br.d, 4 H, *o*-H, 2 Ph, *J* = 7.6 Hz). ¹³C NMR, δ: 21.3 (Me); 73.9 (*p*-TolC≡C); 115.1, 119.6 (*ipso*-C, 2 Ph, *p*-Tol); 130.3, 130.5, 135.1, 136.1, 139.1 (2 Ph, *p*-Tol); 127.2 (*p*-TolC≡C); 142.2 (C≡C, *cyclo*-C₃⁺); 146.5 (CMe, *p*-Tol); 153.9 (2 CPh, *cyclo*-C₃⁺).

Typically, the signals of the β-carbon atom of the triple bond in the ¹³C NMR spectra of perchlorates **3a–c** occur in the region of 126–138 ppm, i.e., in a substantially lower field than the similar signals in the spectra of other acetylene derivatives, for example, *gem*-(alk-1-ynyl)halocyclopropanes (75–95 ppm).^{4,5} This shift is due most likely to the presence of effective conjugation between the triple bond and the cyclopropenium cation in salts **3a–c**.

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